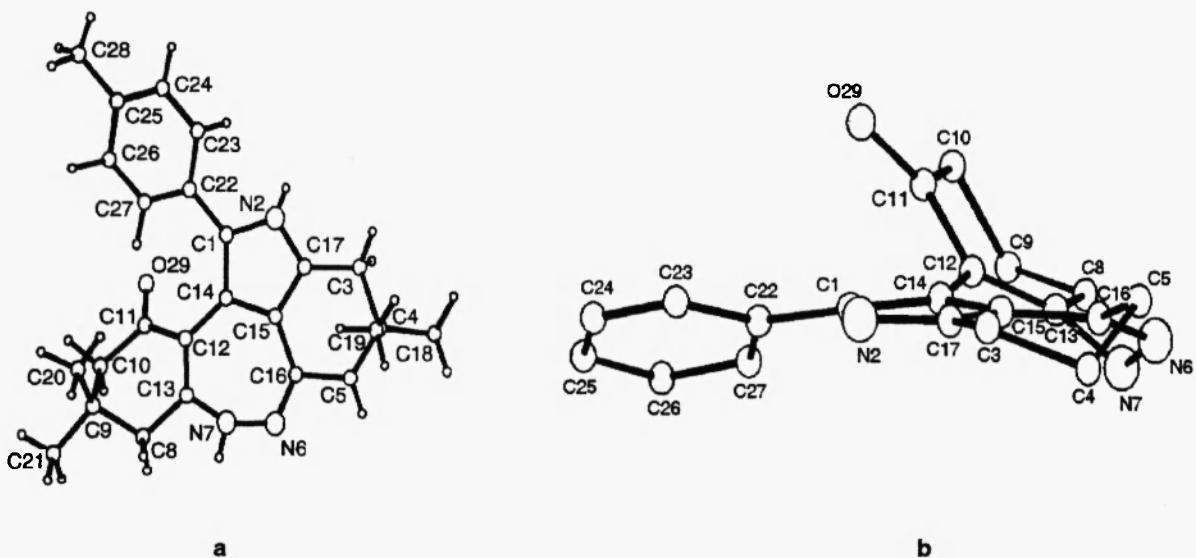


Scheme 1.

X-ray studies

X-ray studies of 4b confirmed the structure proposed on the basis of $^1\text{H-NMR}$ spectral data. Two species of hydrogen bonds of an intermolecular nature are found in the crystal structure of 4b. By shortest of them (N7-H7...N6) two molecules form the centrosymmetric associate in unit cell, the other (N2-H2...O29) bonded those associates parallel to *c* axis of the crystal. The structural fragment of condensed heterocycles is found to be rather planar (angle between plane C1N2C17C15C14 and C14C15C16N6C13C12 is 10°) (Fig.1). Planes of C=O group (O29C11C12) and

Fig.1. Structure of compound 4b: frontal view (a); perspective view with omitted methyl groups and hydrogen atoms (b)

NH group of diazepine ring (N6N7C13) appear out of common conjugated system and, in relation to this one, are oriented on diverse directions. Angles between these planes and plane C14C15C16N6C13C12 are 36° and 48° respectively, nevertheless conjugation of both groups may be assumed. The N2 atom is located at the plane of pyrrole ring. Apparently, NH-group of diazepine ring is unlikely to take part in common conjugation of the system due to considerable bending of these atoms from the plane of seven-membered ring.

Pharmacology

The compound **4a** showed considerably lower toxicity in mice ($LD_{50}>5000$ mg/kg *i.p.*) than **3a** ($LD_{50}>300$ mg/kg *i.p.*).

The compounds weakly influenced the mice behaviour (Table 1): only at high dose of 50 mg/kg **4a** slightly increased the locomotor activity, **3a** at the doses 5 mg/kg and 50 mg/kg induced slight but statistically significant depression of locomotion (actometria test). These compounds exhibited anticonvulsive action in pentylenetetrazole (PTZ) - seizure test at the doses from 0.05 mg/kg to 50 mg/kg, and lengthened the hexobarbital-sleeping time. Compounds **3a** and **4a** had no analgesic (hot plate test) and tranquilizing activity (rota rod test), however **3a** was able to lower body temperature (by at least 3 °C) already from small doses (0.005 mg/kg) to 50 mg/kg. Both compounds (**4a** at the doses 0.5-50 mg/kg) slightly reduced amphetamine-induced hyperactivity indicating anti-dopamine component of their action. No considerable dose-response dependence was observed in tests used.

Experimental protocols

Chemistry

Melting points were determined on a HMR microscope apparatus. Elemental analyses (C,H,N) were within $\pm 0.4\%$ of the theoretical values. UV spectra were recorded on a Hitachi 557 spectrophotometer; peak positions λ_{max} are expressed in nm; log ϵ values are presented in parentheses. IR spectra were recorded on a Perkin Elmer 580B spectrometer, in Nujol; peak positions ν_{max} are expressed in cm^{-1} . $^1\text{H-NMR}$ spectra were recorded on a Bruker WH-90 spectrometer and chemical shifts are reported as δ values (ppm) relative to tetramethylsilane.

1-Phenyl-4,4,9,9-tetramethyl-2,3,4,5,8,9,10,11-octahydro-7H-benzo[*f*][1,2]diazepino[5,4-*c*,*d*]indol-11-one (4a)

A mixture of 5.3 g (0.014 mol) of **2a** and 0.73 ml (0.015 mol) hydrazine monohydrate in 60 ml ethanol was heated in autoclave at 130 °C for 3 h. After cooling the orange crystalline precipitate was filtered and washed with ethanol. Yield - 3 g (57%) of **4a**, m.p. 314-317 °C.

UV ($\text{C}_2\text{H}_5\text{OH}$): 206 (4.33), 230 (4.34), 277 (4.36), shoulder at 350 (3.70), plateau at 400-430 (3.44); IR: 3340, 3290, 3240, 3170, 3100, 3040, 1680, 1625, 1610, 1580; $^1\text{H-NMR}$ (CDCl_3): 1.05 (12H, s, 4-,9- $\text{C}(\text{CH}_3)$), 2.00 (2H, s, 3- CH_2), 2.05 (2H, s, 5- CH_2), 2.10 (2H, s, 8- CH_2), 2.35 (2H, s, 10- CH_2), 5.95 (1H, s, NH), 7.00-7.25 (5H, m, C_6H_5), 7.75 (1H, s, NNH).

Table 1. Neurotopic activity of compounds **4a** and **3a** administered i.p. or i.v. in BALB/C mice 1 hr prior to test.

Compound	Dose, (mg/kg)	Mean \pm SEM						LD ₅₀ (mg/kg)
		Anticonvulsant activity, PTZ convulsions (mg/kg)	Sleeping time (min)	(number of locomotor responses)	Locomo or activity treated mice	Amphetamine treated mice	Body temperature °C	
Control	0	31.6 \pm 0.7	53.2 \pm 7.7	46.6 \pm 1.5	389.5 \pm 16.5	485.6 \pm 30.3	39.2 \pm 0.25	
4a	0.005	30.6 \pm 3.3	74.1 \pm 2.2	53.3 \pm 8.1	377.6 \pm 6.7	434.5 \pm 33.5	39.7 \pm 0.31	>5000
	0.05	31.2 \pm 2.9	96.3 \pm 4.8 ^a	56.5 \pm 2.5 ^a	357.0 \pm 23.3	416.6 \pm 34.5	39.9 \pm 0.51	
	0.5	29.8 \pm 4.4	91.6 \pm 8.6 ^a	68.3 \pm 9.3 ^a	445.3 \pm 17.8	338.5 \pm 25.6 ^a	38.8 \pm 0.48	
	5.0	34.6 \pm 6.2	85.5 \pm 4.5 ^a	59.1 \pm 5.6	450.5 \pm 21.3	340.0 \pm 23.4 ^a	38.6 \pm 0.38	
	50.0	30.2 \pm 1.0	79.4 \pm 2.2 ^a	33.3 \pm 3.1 ^a	737.0 \pm 28.3 ^a	142.5 \pm 18.3 ^a	38.6 \pm 0.72	
Control	0	24.8 \pm 1.4	61.8 \pm 3.1	44.1 \pm 1.1	336.1 \pm 5.5	1054.0 \pm 59.4	39.2 \pm 0.30	
3a	0.005	30.2 \pm 4.8	60.0 \pm 4.8	45.1 \pm 3.4	488.2 \pm 24.4	495.6 \pm 42.3 ^a	36.2 \pm 0.35 ^a	564(342-814)
	0.05	28.6 \pm 3.9	85.1 \pm 3.3 ^a	47.0 \pm 2.2	365.0 \pm 8.9	478.8 \pm 38.8 ^a	36.3 \pm 0.25 ^a	
	0.5	38.0 \pm 2.1 ^a	103.5 \pm 9.4 ^a	51.6 \pm 2.3 ^a	358.5 \pm 20.0	320.8 \pm 23.4 ^a	34.6 \pm 0.37 ^a	
	5.0	41.6 \pm 3.8 ^a	120.1 \pm 6.4 ^a	55.0 \pm 2.8 ^a	272.5 \pm 13.2 ^a	238.8 \pm 18.6 ^a	33.8 \pm 0.53 ^a	
	50.0	42.1 \pm 2.3 ^a	134.0 \pm 12.5 ^a	64.0 \pm 1.5 ^a	224.6 \pm 15.4 ^a	70.5 \pm 8.5 ^a	31.8 \pm 0.48 ^a	

^a p<0.05 vs control (Student's t-test)

1-(4-Tolyl)-4,4,9,9-tetramethyl-2,3,4,5,8,9,10,11-octahydro-7H-benzo[*f*][1,2]diazepino[5,4-c,d]indol-11-one (4b**)**

Compound **4b** was prepared similar to **4a** from 0.3 g (0.8 mmol) of **2b** and 0.04 ml (0.8 mmol) hydrazine monohydrate in 3 ml ethanol. Yield - 0.1 g (32%) of **4b**, m.p. 308-311 °C (dec.).

UV (C₂H₅OH): 204 (4.35), 232 (4.33), 276 (4.37), 356 (3.69), plateau at 400-430 (3.47); IR: 3350, 3300, 3240, 3170, 3100, 3020, 1680, 1630, 1570; ¹H-NMR (CDCl₃): 1.10 (12H, s, 4-,9-C(CH₃)), 2.05 (2H, s, 3-CH₂), 2.10 (2H, s, 5-CH₂), 2.15 (2H, s, 8-CH₂), 2.35 (3H, s, 4'-CH₃), 2.40 (2H, s, 10-CH₂), 5.95 (1H, s, NH), 7.05 (4H, s, C₆H₄), 7.60 (1H, s, NNH).

1-(4-Nitrophenyl)-2,4,4,9,9-pentamethyl-2,3,4,5,8,9,10,11-octahydro-7H-benzo[*f*][1,2]diazepino[5,4-c,d]indol-11-one (4c**)**

A mixture of 1 g (2.3 mmol) of **2c** and 0.11 ml (2.3 mmol) hydrazine monohydrate in 15 ml ethanol was refluxed for 30 min and cooled. The red crystalline precipitate was filtered and washed with ethanol. Yield - 0.4 g (40%) of **4c**; after crystallization from ethanol m.p. 325 °C (dec.).

UV (C₂H₅OH): 203 (4.35), 237 (4.39), shoulder at 272 (4.18), 373 (3.98); IR: 3370, 3300, 3280, 3230, 3170, 3120, 3080, 3040, 1680, 1650, 1620, 1600, 1570; ¹H-NMR (CDCl₃): 1.00 (6H, s, 4-C(CH₃)₂), 1.10 (6H, s, 9-C(CH₃)₂), 1.95 (2H, s, 3-CH₂), 2.10 (2H, s, 5-CH₂), 2.15 (2H, s, 8-CH₂), 2.40 (2H, s, 10-CH₂), 3.25 (3H, s, NCH₃), 6.05 (1H, br.s, NNH), 7.30 and 8.15 (total 4H, AA'BB'-type, m, 2',3',5',6'-H).

X-ray diffraction analysis

The well-shaped yellow crystal of compound **4b** (molecular formula C₂₅H₂₉N₃O) with dimension 0.15×0.25×0.35 mm crystallized from ethanol in the triclinic space group P1 was used for X-ray measurement. The reflection intensities were collected by ω -scan technique at room temperature on the Syntex-P2₁ single-crystal diffractometer using graphite-monochromated Mo-K_α radiation ($\lambda=0.71069$ Å). The cell constants were obtained from a least-squares refinement on the setting angles of 18 reflections in the range of 23°<2θ>30°. The crystallographic data are *a*=7.045(1), *b*=11.314(2), *c*=14.152(2) Å, α =97.23(1), β =100.56(1), γ =103.31(1)°, *V*=1062.3(3) Å³, *Z*=2, *F*(000)=416, *D*_x=1.21 g/cm³, μ (MoK_α)= 0.5 cm⁻¹.

A total of 1984 reflections was collected, of which 1612 were used in calculations. The structure was solved by a direct method using program SHELX86 (6) and refined by full-matrix least squares procedure (program SHELX76 (7)) with the empirical absorption correction (program DIFABS (8)). The positions of hydrogen atoms were found from a difference Fourier map. Non-hydrogen atoms were refined in the anisotropic but hydrogens - in the isotropic approximations. The final reliability factor *R* was 0.0514 (unit weights). The highest peak in the final difference map was 0.20 e/Å³.

Non-hydrogen atomic coordinates and isotropic thermal parameters for atoms of **4b** are given in Table 2 (9).

Acknowledgments

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- (9) **Table 2.** Non-hydrogen atomic coordinates ($\times 10^4$) and isotrophic thermal parameters (esd's in parentheses) for atoms of **4b**.

Atom	x/a	y/b	z/c	B_{eq} (\AA^2)
C1	6566(6)	5921(4)	1223(3)	2.9(2)
N2	7516(6)	4993(4)	1049(3)	3.1(3)
C3	8223(10)	3149(6)	1770(4)	4.1(4)
C4	8461(7)	2841(4)	2810(3)	3.7(2)
C5	6592(8)	2913(5)	3186(4)	3.4(3)
N6	5750(6)	4632(4)	4004(3)	3.4(2)
N7	5522(7)	5849(4)	4119(3)	3.6(2)
C8	3262(9)	7127(5)	3962(4)	3.3(3)
C9	2536(8)	8058(4)	3397(3)	3.7(2)
C10	1580(8)	7377(5)	2356(4)	3.7(2)
C11	2830(7)	6663(4)	1878(4)	3.1(2)
C12	4456(6)	6326(4)	2489(3)	2.7(2)
C13	4450(7)	6393(4)	3458(3)	2.9(2)
C14	5801(6)	5723(4)	2047(3)	2.5(2)
C15	6428(6)	4708(4)	2372(3)	2.7(2)
C16	6190(6)	4175(4)	3218(3)	2.8(2)
C17	7427(7)	4260(4)	1730(3)	3.3(2)
C18	8744(12)	1532(6)	2788(5)	4.9(4)
C19	10273(11)	3742(7)	3475(6)	5.7(3)
C20	4286(11)	9144(5)	3419(5)	4.8(2)
C21	968(13)	8495(8)	3854(6)	5.4(4)
C22	6724(6)	6905(4)	645(3)	2.6(2)
C23	7161(7)	6753(5)	-276(3)	3.2(2)
C24	7415(7)	7709(5)	-788(4)	3.5(3)
C25	7261(7)	8865(5)	-420(3)	3.2(2)
C26	6848(7)	9024(5)	497(4)	3.7(2)
C27	6594(7)	8070(5)	1021(4)	3.5(3)
C28	7596(11)	9924(7)	-971(5)	4.3(3)
O29	2368(5)	6335(3)	990(2)	3.8(1)

$$B_{eq}=8\pi^2 D_u^{1/3}, \text{ where } D_u \text{ is the determinant of the U matrix in orthogonal space}$$

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